

# Synthesis and Steric Structure of [2-(3-Pyridyl)piperidino](2-vinyloxyethylamino)methanethione

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**Abstract**—Thioureas derivatives were synthesized on the basis of anabasine, piperidine, and morpholine, and 2-vinyloxyethyl isothiocyanate. X-Ray diffraction was used to determine the steric structure of [2-(3-pyridyl)piperidino](2-vinyloxyethylamino)methanethione.

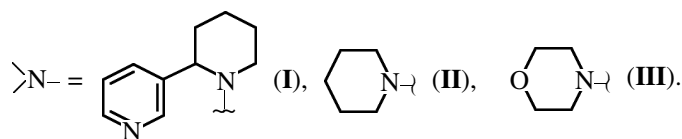
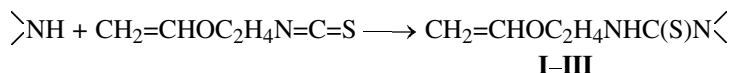
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The interest in thiourea and its derivatives is explained by their diverse biologic activity. Specifically, these compounds have exhibited valuable pharmacologic properties [1], as well as insecticide [2], fungicide [3], and antitubercular activity [4].

Previously we made use of the reaction of anabasine with dialkoxyposphoryl isothiocyanates to synthesize a series of thioureas that exhibited high

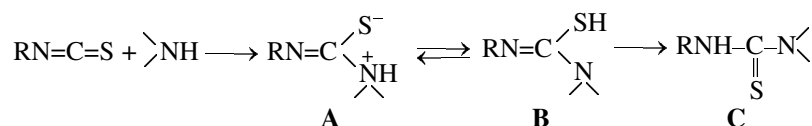
antimicrobial and insecticide activity and low toxicity in screening tests [5, 6]. To reveal structure–bioactivity interrelationship in this series of compounds, we set ourselves the task to synthesize new thiourea derivatives on the basis of mentioned compounds.

The condensation of 2-vinyloxyethyl isothiocyanate with heterocyclic amines was performed in alcohols or acetone at equimolar reactant ratios.



The addition of amines to isothiocyanates classed with heterocumulenes presumably occurs by the following scheme: Nucleophile (amine with a lone electron pair) attacks the electron-deficient center in iso-

thiocyanate, viz. the  $\text{N}=\text{C}=\text{S}$  carbon atom, to form dipolar intermediate **A**. Proton transfer from nitrogen to sulfur gives rise to isothioureia **B** that readily isomerizes into a more stable form, thiourea **C**.



The reaction products are colorless crystals (**I**, **III**) or a light yellow oil (**II**). Their composition, structure, and individuality were proved by elemental analysis,  $^1\text{H}$  NMR and IR spectroscopy, and TLC. The IR spectra of compounds **I–III** contain  $\text{C}=\text{S}$  and  $\text{C}-\text{O}-\text{C}$  absorption bands at  $1550\text{--}1525$  and  $1205\text{--}1190\text{ cm}^{-1}$ , and the  $\text{C}=\text{C}$  bond appears at  $1635\text{--}1620\text{ cm}^{-1}$ .

To establish the steric structure of [2-(3-pyridyl)-piperidino](2-vinyloxyethylamino)methanethione (**I**), we made use of X-ray diffraction analysis. The general view of molecule **I** is given in the figure.

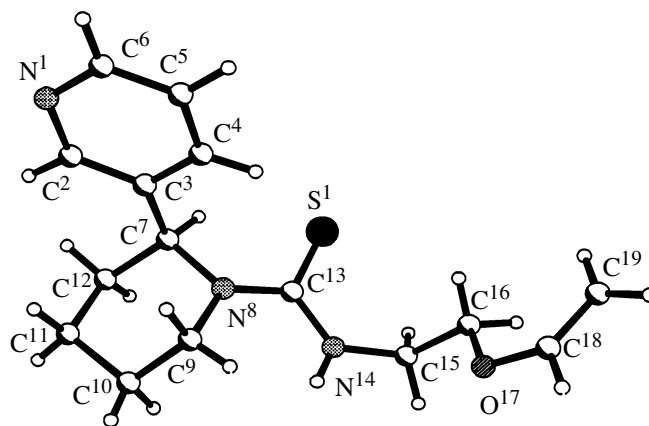
The bond lengths and angles are close to normal. The piperidine ring adopts a *chair* conformation ( $\Delta C_s^8$  2.2 Å), like in *O,O*-diethyl (3-pyridyl)piperidino thiophosphate (**IV**) and *O,O*-diisopropyl (3-pyridyl)piperidino thiophosphate (**V**) [7]. The pyridine ring is planar within  $\pm 0.009$  Å. The pyridine ring is arranged axially ( $\text{C}^3\text{C}^7\text{N}^8\text{C}^9$  torsion angle  $72.4^\circ$ ) with respect to the piperidine ring, which distinguishes compound **I** from **IV** and **V**, where this ring is equatorial. In our opinion, this is explained by the nonbonded repulsion between the bulky substituent on  $\text{N}^8$  and the pyridine ring.

## EXPERIMENTAL

The IR spectra were measured on an Avatar-320 instrument in KBr pellets and thin films. The  $^1\text{H}$  NMR spectra of  $\text{C}_6\text{D}_6$  solutions were recorded on a Varian Mercury-300 instrument at 300 MHz. The melting points were measured on a Boetius hot stage.

**X-ray diffraction experiment.** The unit cell parameters and the intensities of 1721 unique reflections were measured on a Syntex-P21 diffractometer ( $\text{CuK}\alpha$  radiation, graphite monochromator,  $\theta/2\alpha$  scanning,  $2\theta \leq 138.8^\circ$ ). Rhombic crystals,  $a$  7.509(2),  $b$  13.461(3),  $c$  15.667(3) Å;  $V$  1583.6(6) Å<sup>3</sup>,  $d_{\text{calc}}$  1.222 g cm<sup>-3</sup>,  $Z$  4.  $\text{C}_{15}\text{H}_{21}\text{N}_3\text{OS}$ . Space group  $P2_12_12_1$ . The structure was solved by the direct method and refined by full-matrix least squares in the anisotropic approximation for nonhydrogen atoms. Hydrogen atoms were revealed by difference synthesis in the anisotropic approximation. The calculations involved 1641 reflections with  $I > 2\sigma(I)$ . Final divergence factors:  $R$  0.037 and  $wR_2$  0.098. The structure was solved and refined using the SHELXS-97 and SHELXL-97 programs. The atomic coordinates are available from the authors.

**[2-(3-Pyridyl)piperidino](2-vinyloxyethylamino)-methanethione (I).** To a solution of 16.7 g of anabasine in 20 ml of benzene, a solution of 12.9 g of 2-vinyloxyethyl isothiocyanate in 25 ml of absolute benzene was added dropwise with stirring at room



Molecular structure of [2-(3-pyridyl)piperidino](2-vinyloxyethylamino)methanethione (**I**).

temperature. The reaction mixture was then stirred at room temperature for 1 h and at  $35\text{--}40^\circ\text{C}$  for 2 h, and the solvent was distilled off. Yield 75–83%.  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 8.40 d (1H,  $\text{CH}^a$ ,  $^3J$  4.8 Hz), 7.09 d.d (1H,  $\text{CH}^b$ ,  $^3J$  7.4 Hz), 7.52 d (1H,  $\text{CH}^c$ ), 1.32–1.58 m (6H, ring  $\text{CH}_2$ ), 3.16 t (2H,  $\text{NCH}_2$ ), 3.42 t (2H,  $\text{CH}_2\text{O}$ ), 6.35 q (1H,  $\text{CHO}$ ), 3.86 d.d (2H,  $\text{CH}=\text{CH}_2$ ,  $J$  5.7 Hz). Found, %: C 61.87; H 7.19; N 14.49.  $\text{C}_{15}\text{H}_{21}\text{N}_3\text{OS}$ . Calculated, %: C 61.82; H 7.26; N 14.42.

**Piperidino(2-vinyloxyethylamino)methanethione (II)** was prepared in a similar way as a light yellow oily material. Yield 87–90%.  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 1.57–2.12 m (10H, ring  $\text{CH}_2$ ), 3.21 t (2H,  $\text{NCH}_2$ ), 3.44 t (2H,  $\text{CH}_2\text{O}$ ), 6.37 q (1H,  $\text{CHO}$ ), 3.89 d.d (2H,  $\text{CH}=\text{CH}_2$ ,  $J$  5.6 Hz). Found, %: C 56.11; H 8.41; N 13.15.  $\text{C}_{10}\text{H}_{18}\text{N}_2\text{OS}$ . Calculated, %: C 56.04; H 8.47; N 13.07.

**Morpholino(2-vinyloxyethylamino)methanethione (III)** was prepared in a similar way as colorless crystals. Yield 79–83%, mp  $56\text{--}58^\circ\text{C}$ .  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 2.98 m (4H,  $\text{NCH}_2$ ), 3.15 m (4H,  $\text{OCH}_2$ ), 3.19 t (2H,  $\text{NCH}_2$ ), 3.40 t (2H,  $\text{CH}_2\text{O}$ ), 6.38 q (1H,  $\text{CHO}$ ), 3.87 d.d (2H,  $\text{CH}=\text{CH}_2$ ,  $J$  5.6 Hz). Found, %: C 50.04; H 7.38; N 12.99.  $\text{C}_9\text{H}_{16}\text{N}_2\text{O}_2\text{S}$ . Calculated, %: C 49.98; H 7.46; N 12.95.

## REFERENCES

1. Mashkovskii, M.D., *Lekarstvennye sredstva* (Drugs), Moscow: Meditsina, 1993, Vols. 1, 2.
2. Mel'nikov, N.N., Volkov, A.I., and Korotkova, O.A., *Pestitsidy i okruzhayushchaya sreda* (Pesticides and Environment), Moscow: Khimiya, 1977.

3. Mozolis, V.V. and Iokubaitite, S.P., *Usp. Khim.*, 1973, vol. 42, no. 7, p. 1310.
4. Zhangereev, A., *Densaulyk saktau Kazahstan*, 1998, nos. 7–8, p. 2.
5. Fazylov, S.D., *Doctoral (Chem.) Dissertation*, Karaganda, 1999.
6. Zhivotova, T.S., *Cand. Sci. (Chem.) Dissertation*, Karaganda, 1999.
7. Gazaliev, A.M., Zhurinov, M.Zh., Dyusembaev, S.A., Turdybekov, K.M., Lindeman, S.V., and Struchkov, Yu.V., *Khim. Geterotsikl. Soedin.*, 1990, no. 4, p. 500.